

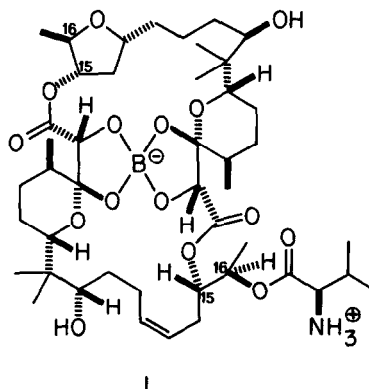
(2S,3R)-1,2-EPOXY-3-BUTANOL. A USEFUL SYNTHON  
FOR THE PREPARATION OF CHIRAL 1,2-DIOLS.

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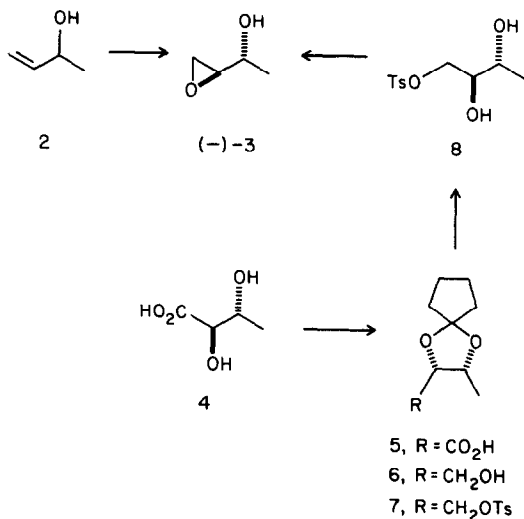
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**Summary:** Enantioselective epoxidation of 3-butene-2-ol in the presence of D-(-)-tartrate gave (2S,3R)-1,2-epoxy-3-butanol, which was employed in chiral syntheses of 2,5-dideoxyribose and a segment of the ionophoric antibiotic boromycin.

The enantioselective epoxidation of racemic allylic alcohols mediated by optically active tartrate esters represents a powerful method for the synthesis of chiral, polyhydroxylic systems.<sup>1</sup> In connection with studies directed towards the total synthesis of boromycin (1),<sup>2</sup> we required a synthon incorporating the C(15,16) diol function of this antibiotic which possessed the natural (15S,16R) configuration. We describe herein the asymmetric synthesis of (2S,3R)-1,2-epoxy-3-butanol (3), an independent proof of its configuration, and its transformation to 2,5-dideoxyribose and to a segment corresponding to C(11)-C(17) of 1.



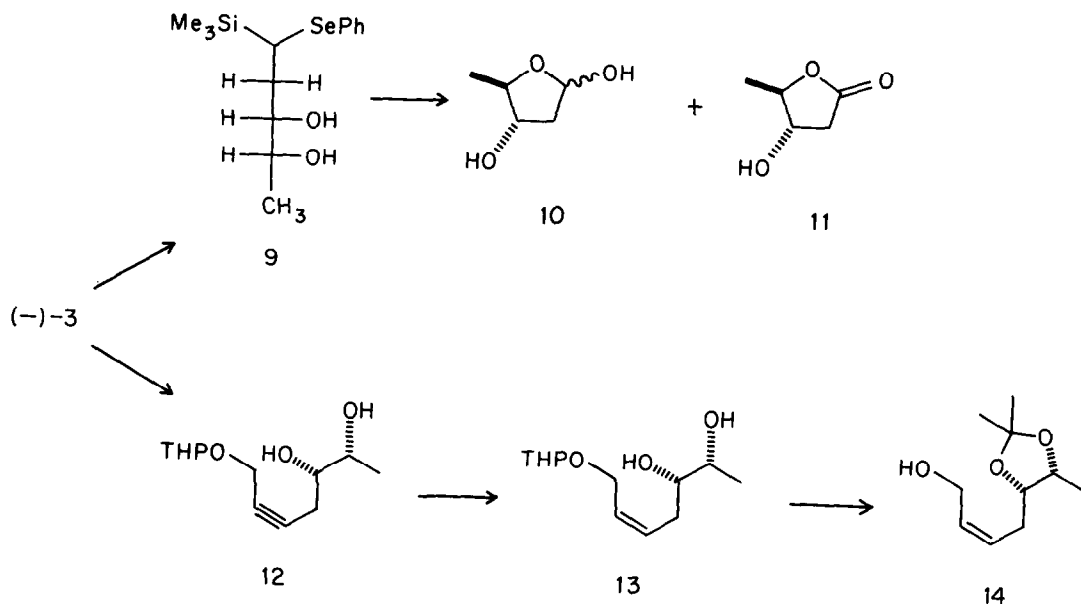
Epoxidation of 3-butene-2-ol (2) with *t*-butyl hydroperoxide and titanium tetraisopropoxide in the presence of D-(-)-diisopropyl tartrate, followed by distillation, afforded 3 [bp 50°C, 15 mm Hg;  $[\alpha]_D^{22} - 16.3^\circ$  ( $c$  0.97, MeOH)] in 54% yield. Although precedence suggested that the 3R enantiomer of 2 should be the faster-reacting antipode,<sup>1</sup> yielding predominantly the erythro epoxide in which attack had occurred selectively from the si face,<sup>3</sup> an unambiguous proof of this stereochemistry was desired.



To this end, (±)-erythro-2,3-dihydroxybutyric acid (4), prepared by hydroxylation of *trans*-crotonic acid,<sup>4</sup> was resolved via its quinine salt.<sup>5,6</sup> Optical rotatory dispersion studies have established that (-)-4 possesses (2R,3R) absolute configuration.<sup>7</sup> Protection of (-)-4 as its ketal derivative 5 (91%) with cyclopentanone and a catalytic quantity of *p*-toluenesulfonic acid in benzene, followed by reduction with lithium aluminum hydride in THF gave the primary alcohol 6 in 87% yield. Conversion of this substance to its tosylate 7 (95%; mp 41-43°C) with *p*-toluenesulfonyl chloride in pyridine and removal of the cyclopentylidene residue with methanol and a catalytic quantity of *p*-toluenesulfonic acid gave the diol 8 (73%; mp 69-71°C). Treatment of 8 with sodium hydride in THF containing a trace of DMSO produced 3 [72%;  $[\alpha]_D^{20} - 17.9^\circ$  ( $c$  1.16, MeOH)],<sup>8</sup> which was identical by both spectroscopic and gas chromatographic comparison with the material obtained from 2. From this result it was ascertained that asymmetric epoxidation of 2 had afforded a 91% enantiomeric excess of the (2S,3R) epoxy alcohol 3.

The utility of 3 as a precursor to chiral 1,2-diols was first examined in the context of a deoxypentose synthesis, where alkylation at the epoxide terminus with a formyl anion equivalent was required. The reaction of (-)-3 with

the anion of phenylselenomethyltrimethylsilane<sup>9</sup> (2 equiv LDA, THF, -78°C to +25°C, 12h) gave **9** as a C(1) epimeric mixture in 80% yield. This mixture was subject-



ed to a Pummerer rearrangement with 30% H<sub>2</sub>O<sub>2</sub> in THF-ether (4:6) to yield 2,5-dideoxyribofuranose (**10**, 40%),<sup>10</sup> accompanied by the deoxyribonolactone **11** [24%,  $[\alpha]_D^{18} + 21.0^\circ$  ( $c$  0.67, CHCl<sub>3</sub>)].

For construction of the erythro-dihydroxyheptenyl segment, corresponding to C(11)-C(17) of **1**, alkylation of (-)-**3** was effected with the anion of propargyl tetrahydropyran ether, prepared with *n*-butyllithium in THF, to furnish **12** in 90% yield. Semihydrogenation of the acetylenic linkage over 10% Pd/BaSO<sub>4</sub> in the presence of quinoline afforded a 98% yield of the cis olefin **13** as an epimeric mixture at the tetrahydropyran center. Treatment of **13** with methanol containing *p*-toluenesulfonic acid and then, following evaporation of the volatiles, with acetone-benzene (1:1), gave **14** [ $[\alpha]_D^{20} - 22.5^\circ$  ( $c$  4.92, CHCl<sub>3</sub>)] in 74% yield after chromatography.

This substance embodies the functionality and absolute configuration of an important structural component of **1** and provides a convenient starting point for elaborating the "southern" half of this macrodiolide.

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## REFERENCES

- (1) Martin, V.S.; Woodard, S.S.; Katsuki, T.; Yamada, Y.; Ikeda, M.; Sharpless, K.B. J. Am. Chem. Soc. 1981, 103, 6237.
- (2) White, J.D.; Avery, M.A.; Choudhry, S.C.; Dhingra, O.P.; Kang, M.-C.; Sheldon, B.G.; Whittle, A.J. J. Am. Chem. Soc. submitted for publication.
- (3) Katsuki, T.; Sharpless, K.B. J. Am. Chem. Soc. 1980, 102, 5974.
- (4) Mugdan, M.; Young, D.P. J. Chem. Soc. 1949, 2988.
- (5) Hoff-Jorgensen, E. Z. Physiol. Chem. 1941, 268, 194.
- (6) (-)-Erythro-2,3-dihydroxybutyric acid showed  $[\alpha]_D^{20} - 10.78^\circ$  (c 1.17, H<sub>2</sub>O) [lit<sup>7</sup>  $[\alpha]_D^{20} - 9.50^\circ$  (c 1.0, H<sub>2</sub>O)].
- (7) Bachelor, F.W.; Miana, G.A. Can. J. Chem. 1969, 47, 4089.
- (8) <sup>1</sup>H NMR of 3:  $\delta$  (CDCl<sub>3</sub>) 3.96 (d of q, J=7,3 Hz, 1H), 3.00 (q, J=3 Hz, 1H), 2.78 (bd, J=3 Hz, 2H), 2.30 (bs, 1H, exchanged with D<sub>2</sub>O), and 1.23 (d, J=7 Hz, 3H).
- (9) Sachdev, J.; Sachdev, H.S. Tetrahedron Lett. 1976, 4223.
- (10) Zinner, H.; Wigert, H. Chem. Ber. 1959, 92, 2893.

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